

# Assignment of the $Q_y$ Absorbance Bands of Photosystem II Chromophores by Low-Temperature Optical Spectroscopy of Wild-Type and Mutant Reaction Centers<sup>†</sup>

David H. Stewart,<sup>‡,§</sup> Peter J. Nixon,<sup>||</sup> Bruce A. Diner,<sup>⊥</sup> and Gary W. Brudvig<sup>\*,‡</sup>

*Sterling Chemistry Laboratory, Yale University, P.O. Box 208107, New Haven, Connecticut 06520-8107, Central Research and Development Department, Experimental Station, E. I. du Pont de Nemours and Company, Wilmington, Delaware 19880, and Department of Biochemistry, Imperial College, London, SW7 2AY, United Kingdom*

Received May 31, 2000

**ABSTRACT:** Photosystem II (PSII) contains a collection of pheophytins (Pheo) and chlorophylls (Chl) that have unique absorbance spectra depending on their electronic structure and the surrounding protein environment. Despite numerous efforts to identify the spectra of each cofactor, differing assignments of the chromophore absorbance bands and electrochromic effects have led to conflicting models of pigment organization and chromophore interactions in PSII. We have utilized low-temperature measurements on well-defined redox states, together with the use of site-directed mutants, to make spectral assignments of several reaction center (RC) chromophores. Cryogenic (77 K) optical spectroscopy has been used to trap the bound redox-active quinone,  $Q_A$ , in the reduced form and measure the effect of the redox state of  $Q_A$  on PSII chromophores without interference from other redox-active cofactors. The  $Q_A^-$  minus  $Q_A$  difference spectrum contains a number of features that represent the perturbation of Pheo and Chl absorbance bands upon  $Q_A$  reduction. Using site-directed mutants in which the axial ligand of the D1-side monomeric core Chl,  $P_A$ , is changed (D1-H198Q) or the hydrogen-bonding environment of the D1-side Pheo is modified (D1-Q130E), we have assigned the  $Q_y$  absorbance bands of four chromophores shifted by  $Q_A$  reduction including both RC Pheos, the D1-side monomeric accessory Chl ( $B_A$ ), and one other Chl in PSII. The absorbance maximum of  $B_A$  was identified at 683.5 nm from least-squares fits of the D1-H198Q minus wild type (WT)  $Q_A^-$  minus  $Q_A$  double-difference spectrum; this assignment provides new evidence of a secondary effect of site-directed mutation on a RC chromophore. The other chromophores were assigned from simultaneous fits of the WT and D1-Q130E spectra in which the parameters of only the D1-side Pheo were allowed to vary. The D1-side and D2-side Pheos were found to have  $\lambda_{\max}$  values at 685.6 and 669.3 nm, respectively, and another Chl influenced by  $Q_A^-$  was identified at 678.8 nm. These assignments are in good agreement with previous spectral analyses of intact PSII preparations and reveal that the number of chromophores affected by  $Q_A$  reduction has been underestimated previously. In addition, the assignments are generally consistent with chromophore positions that are similar in the PSII RC and the bacterial photosynthetic RC.

Photosystem II (PSII)<sup>1</sup> is a membrane-bound pigment–protein complex that converts light energy into chemical energy in order to oxidize  $H_2O$  to  $O_2$ . The minimal structure required for photochemistry—the reaction center (RC)—is comprised of the D1 and D2 polypeptides, cytochrome  $b_{559}$  (cyt  $b_{559}$ ), and the PsbI protein and contains six chlorophyll (Chl) molecules and two pheophytin (Pheo) molecules (1–

6). The PSII core complex includes the RC plus two Chl-binding proteins (CP43, CP47), extrinsic polypeptides bound outside the membrane, and several low molecular mass subunits (7). When illuminated with visible light, Chls in PSII absorb and transfer the light energy to a Chl molecule(s), P680, which initiates photochemistry (reviewed in ref 8). The excited state of P680 (P680\*) donates an electron to a

<sup>†</sup> This work was supported by Grant GM32715 (G.W.B.) from the National Institutes of Health, Grants 96-35306-3398 (G.W.B.) and 94-37306-0783 (B.A.D.) from the National Research Initiative Competitive Grants Program/USDA, by a National Institutes of Health predoctoral traineeship, GM08283 (D.H.S.), and The Royal Society and BBSRC (P.J.N.).

\* To whom correspondence should be addressed. Telephone: (203)-432-5202. Fax: (203)432-6144. E-mail: gary.brudvig@yale.edu.

<sup>‡</sup> Yale University.

<sup>§</sup> Present address: Xantho, Inc., P.O. Box 12296, Research Triangle Park, NC 27709.

<sup>||</sup> Imperial College.

<sup>⊥</sup> E. I. du Pont de Nemours and Company.

<sup>1</sup> Abbreviations:  $B_A/B_B$ , monomeric accessory Chls on the D1/D2 sides of PSII; BChl, bacteriochlorophyll; BPheo, bacteriopheophytin; BRC, non-sulfur purple bacterial photosynthetic reaction center; Car, redox-active carotenoid in PSII; Chl, chlorophyll; Chl<sub>D</sub>, accessory Chl bound to D2-H117; Chl<sub>Z</sub>, accessory Chl bound to D1-H118; CP43/CP47, Chl-binding proteins in PSII; cyt  $b_{559}$ , cytochrome  $b_{559}$ ; D1/D2 polypeptides, reaction center polypeptides in PSII; EPR, electron paramagnetic resonance; LT, low temperature; L/M subunits, reaction center polypeptides in the BRC; MQ, menaquinone; P680, primary electron donor in PSII;  $P_A/P_B$ , monomeric reaction center core Chls on the D1/D2 sides of PSII; Pheo, pheophytin; PSI, photosystem I; PSII, photosystem II;  $Q_A/Q_B$ , redox-active quinones in PSII; RC, reaction center; RT, room temperature; WT, wild type;  $Y_Z/Y_D$ , redox-active tyrosines D1-Y161/D2-Y160 in PSII of *Synechocystis* PCC 6803.

Pheo acceptor to create a charge-separated state,  $\text{P680}^+ \text{Pheo}^-$ . The charge separation is stabilized on the acceptor side by electron transfer from  $\text{Pheo}^-$  to a bound quinone,  $\text{Q}_\text{A}$ , and on the donor side by electron transfer from a redox-active tyrosine,  $\text{Y}_\text{Z}$ , to  $\text{P680}^+$ . Reduction of  $\text{Y}_\text{Z}^*$  is accomplished by a  $\text{Mn}_4$  cluster that builds up oxidizing equivalents during subsequent charge separations in order to split  $\text{H}_2\text{O}$ .  $\text{Q}_\text{A}^-$  is oxidized by an exchangeable quinone that binds in a second quinone ( $\text{Q}_\text{B}$ ) binding site.

Because the molecular structure of PSII has not been determined at the atomic level, structural considerations of PSII have been based on analogies to the crystal structure of the non-sulfur purple bacterial photosynthetic reaction center (BRC) (9–11). These comparisons are strongly supported by recent electron crystallography of the PSII CP47 RC at a resolution of 8 Å, which indicates that the positions of both the polypeptides and the chlorin cofactors of PSII are very similar to those in the BRC although the pair of core Chls analogous to the special pair bacteriochlorophyll (BChl) dimer are separated by a greater distance (~11 Å versus 7.6 Å) (12). In addition, the BRC and the PSII RC are closely related by significant protein sequence similarity (reviewed in ref 9), 2-fold symmetry (9, 10, 13, 14), and similar electron-transfer pathways (15). For example, the two histidines in the L and M subunits of the BRC that bind the special pair BChl dimer (L-H173 and M-H200) are conserved in the D1 and D2 subunits of the PSII RC (D1-H198 and D2-H197) (9). In PSII, these histidines coordinate two core Chls,  $\text{P}_\text{A}$  and  $\text{P}_\text{B}$ , which are the likely site of P680 (9). Two other BChls are bound as monomeric accessory (voyeur) BChls and are ligated by L-H153 and M-H180 in the BRC (16). Although these histidines do not have direct homologues in the PSII RC (9), there are several potential Chl ligands in the D1 and D2 sequences that would allow binding of two voyeur-like Chls ( $\text{B}_\text{A}$  and  $\text{B}_\text{B}$ ) in PSII (14), a scenario that is likely given the homology of the charge-separation reactions of the BRC and PSII (17). Because the BRC binds only four BChls and two bacteriopheophytins (BPheos), there are two Chls in the PSII RC that do not have homologues in the BRC. These Chls have been identified as the monomeric accessory Chls,  $\text{Chl}_\text{Z}$  and  $\text{Chl}_\text{D}$ , and are bound to histidines D1-H118 and D2-H117, which also lack homologues in the BRC (14).

The dense collection of chromophores and redox-active cofactors in the PSII RC creates an environment in which electron transfer through the protein can induce spectral shifts of some of the chromophores. This phenomenon of induced band shifts occurs in a manner that is dependent on the change in charge, the influence of this charge on the surrounding protein matrix, the dipole orientation and H-bonding of the oxidized/reduced cofactor and chromophores, and the spatial relationship of the chromophores relative to the oxidized/reduced cofactor (18). When the band shift is strictly the result of the chromophore interacting with a charged species, it is considered an electrochromic band shift and depends on the orientation of the chromophore, the distance between the charge and the chromophore, and the intervening dielectric of the protein medium between the charge and the chromophore (19). While redox-induced chromophore band shifts complicate the overall spectral properties of PSII, they can be exploited to obtain information about the identity of the individual chromophores. In

particular, difference spectra can be used to great advantage to assign the spectra of individual chromophores contributing to the composite PSII spectrum.

Over the last three decades, there have been many studies aimed at identifying the Chls and Pheos in PSII and assigning the individual absorbance bands with which they are associated. Some approaches have utilized time resolution, chemical treatment, or LT to isolate and probe absorbance changes associated with different redox states of PSII. Several recent efforts include simulations of time-resolved  $\text{P680}^+ \text{Pheo}^- / \text{P680Pheo}$  and  $\text{P680}^+ \text{Q}_\text{A}^- / \text{P680Q}_\text{A}$  spectra (20, 21), characterization of the effect of Pheo oxidation state on PSII spectra (22), and simulations of multiple time-resolved spectra from different redox changes in PSII to predict chromophore topology (23). Another strategy has been to utilize simulations of PSII spectra as a function of temperature in order to deconvolute the contribution from each chromophore (24). There have also been efforts to characterize PSII preparations in which the Chl content has been systematically changed by different detergent treatments or isolation procedures (5, 6, 25, 26). While significant progress has been made, a consensus on definitive chromophore assignments remains elusive because of the spectral congestion in the PSII RC and the difficulty of preparing a sample with a well-defined, homogeneous redox state.

In this paper, we describe an alternative approach to assigning the absorbance bands of PSII chromophores that combines the use of site-directed mutants in which chromophore environments are perturbed with optical characterization of a well-defined redox state change of PSII. A series of mutations were made in close proximity to chromophores in the RC including the D1-side monomeric core Chl ( $\text{P}_\text{A}$ ), the D1-side Pheo ( $\text{Pheo}(\text{D1})$ ), and the accessory Chls ( $\text{Chl}_\text{Z}$  and  $\text{Chl}_\text{D}$ ). Utilizing a method for obtaining a  $\text{Q}_\text{A}^-$  minus  $\text{Q}_\text{A}$  spectrum free of interference from other redox changes, the spectral signatures of this redox state change were identified in the green and red wavelength regions in Mn-depleted PSII core complex preparations. In particular, this method of chemical treatment and LT trapping enabled isolation of the redox change of a single cofactor. Least-squares fits of the  $\text{Q}_\text{A}^-$  minus  $\text{Q}_\text{A}$  spectra were used to identify the features contributing to each spectrum, and assignments of the chromophores contributing to the spectra were made based on comparisons between mutant and wild-type (WT) spectra. Besides providing a specific probe of individual chromophores in the RC, this approach offers information about the topology of cofactors and chromophores in PSII. Insights obtained from this work are discussed in light of previous structural and spectroscopic chromophore assignments in PSII, as well as the BRC structure.

## MATERIALS AND METHODS

**Strains and Construction of Mutants.** Site-directed mutations were introduced into a glucose-tolerant strain of the cyanobacterium, *Synechocystis* PCC 6803 (27). The D1-Q130E, D1-H198Q, and D1-H118Q/D2-H117Q mutants were prepared as described in refs 28, 29 and 14, respectively. All mutations were confirmed by sequencing PCR products amplified from fully segregated *Synechocystis* mutants.

**PSII Isolation and Preparation.** WT and mutant PSII core complexes comprising the D1 and D2 polypeptides, cyt  $b_{559}$ ,

CP47, CP43, and 33-kDa protein and containing about 40 Chl per PSII were purified from *Synechocystis* according to the method of Tang and Diner (30). Contaminating photosystem I (PSI) was at a low level as determined by electron paramagnetic resonance (EPR) and sodium dodecyl sulfate–polyacrylamide gel electrophoresis, and typical  $O_2$ -evolution rates were  $2100\text{--}3200\ \mu\text{mol of } O_2\ (\text{mg of Chl})^{-1}\ \text{h}^{-1}$ . The core complexes were further simplified by removal of the  $Mn_4$  cluster with 5 mM hydroxylamine treatment and washing with 5 mM ethylenediaminetetraacetic acid in pH 6.5 buffer (31). Chl *a* concentrations were determined by methanol extraction and using an extinction coefficient of  $79.24\ \text{mL}\ (\text{mg of Chl})^{-1}\ \text{cm}^{-1}$  at 665 nm (32). All protein purification and manipulation was carried out at 4 °C under dim green light. Sample concentrations were typically  $\sim 150\ \mu\text{g of Chl/mL}$  ( $\sim 4.4\ \mu\text{M}$  PSII) for optical experiments and 0.27 mg of Chl/mL for EPR experiments.

**Preparation of Redox States.**  $Q_A^-$  was generated by room temperature (RT) illumination in the presence of the exogenous electron donor,  $NH_2OH$ , in a series of illuminations that allowed for the collection of spectra from PSII in several redox states. Prior to  $NH_2OH$  treatment and illumination, the PSII was dark-adapted on ice for  $\geq 1\ \text{h}$  to allow for dark reduction of  $Y_D^\bullet$  (33, D. H. Stewart and G. W. Brudvig, unpublished results) so that the redox state of  $Y_D$  would not change during the course of the experiment due to slow reduction of  $Y_D^\bullet$  by  $NH_2OH$ . Afterward, the PSII was dark-adapted on ice (3–5 min) with 1 equiv of  $K_3Fe(CN)_6$  to oxidize dark-reduced  $Q_A$  and cyt  $b_{559}$  while avoiding a significant amount of residual oxidant, and then a 1000-fold excess of  $NH_2OH$  was added to the sample in the dark. Following the addition of  $NH_2OH$ , the sample was frozen to 77 K within 2 min to avoid chemical reduction of  $Q_A$  and cyt  $b_{559}$  by  $NH_2OH$  prior to collection of the nonilluminated scans. After collection of the nonilluminated scans at 77 K, the sample was illuminated at 77 K for 4–8 min to generate the  $Chl_Z^+/Car^+/Q_A^-$  state, and 77 K illuminated scans were collected. Then the sample was thawed to RT in the dark over 2–3 min, allowing for reoxidation of  $Q_A$  by charge recombination. Next, the  $Q_A^-$  state was trapped by illuminating for 1 s at RT, dark adapting for 1 s at RT (to allow rereduction of  $Chl_Z^+$ ,  $Car^+$ , and  $Y_Z^\bullet$ ), and rapid freezing to 77 K in the dark. The freezing procedure involved immediately placing the sample in the cold vapor directly above ( $< 1\ \text{cm}$ ) a liquid  $N_2$  bath for 30 s and then gradually immersing the sample into the liquid  $N_2$  over a 30-s period. After spectra of the RT illuminated sample were collected, the sample was illuminated for 4–8 min at 77 K, and 77 K illuminated spectra were collected to determine the amount of  $Q_A$  still oxidized after the RT illumination (see Results for an explanation of this analysis). In PSII core complexes, it is not necessary to add 3-(3,4-dichlorophenyl)-1,1-dimethylurea (DCMU) to block electron transfer beyond  $Q_A$  at RT because  $Q_B$  is absent from these preparations. The yields of  $Chl_Z^+/Car^+$  and  $Q_A^-$  were measured by EPR and optical spectroscopy. The levels of  $Y_D^\bullet$ ,  $Y_Z^\bullet$ , and  $P700^+$  contamination in all of the difference spectra were less than 5%. All illuminations were performed using an Oriel quartz-halogen lamp ( $700\ \text{W/m}^2$ ). The cycle of freezing, thawing, and refreezing the sample had no effect on the absorbance spectrum of PSII in the absence of actinic illumination.

For comparison to the  $Q_A^-$  state, the charge separated state  $Chl_Z^+/Car^+/Q_A^-$  was formed by continuous illumination of PSII with white light for 4–8 min (depending on the optical density of the sample) in a liquid nitrogen bath. Pretreatment with a 20-fold excess of  $K_3Fe(CN)_6$  fully oxidized  $Q_A$  and cyt  $b_{559}$ , ensuring that all PSII centers were capable of forming reduced  $Q_A$  upon illumination and that  $Chl_Z^+/Car^+$  would not be rereduced by cyt  $b_{559}$  during the illumination. For the purpose of EPR quantitation,  $Y_D^\bullet$  minus  $Y_D$  (illuminated minus nonilluminated) difference spectra were produced by collecting nonilluminated scans of PSII that had been dark-adapted for  $\geq 1.5\ \text{h}$  in the absence of  $K_3Fe(CN)_6$  and then dark-adapted for  $\geq 30\ \text{min}$  in the presence of a 20-fold excess of  $K_3Fe(CN)_6$  (to allow for dark reduction of  $Y_D^\bullet$  and then dark oxidation of  $Q_A^-$  and cyt  $b_{559}$ , respectively). Illuminated scans were collected after illumination at RT for 30 s, dark adaptation at RT for 30 s to allow for the rereduction of electron donors other than  $Y_D$  ( $Y_Z^\bullet$ ,  $Chl_Z^+$ , and  $Car^+$ ), and freezing to 77 K.

**LT Optical Spectroscopy.** Optical spectra were collected using a home-built Plexiglas flat cell submerged in a liquid  $N_2$  bath in a quartz finger Dewar (Wilma WG-850-B-Q). The flat cell had a  $\sim 0.75\ \text{mm}$  path length and a  $\sim 45\text{-}\mu\text{L}$  sample volume. The unsilvered finger section of the Dewar was aligned in the sample beam of a Perkin-Elmer Lambda 6 or Lambda 20 UV–Vis spectrophotometer, and the flat cell was held in place by a plastic collar with a diameter slightly less than the inner diameter of the Dewar ( $\sim 5\ \text{mm}$ ). Interference from bubbles generated by liquid  $N_2$  boil-off was avoided by careful cleaning of the Dewar with basic ethanol (0.125 g of KOH/mL) to limit nucleation sites on the quartz surface. Interference from frozen condensed moisture was minimized by maintaining a dry  $N_2$  atmosphere above the Dewar. Before insertion into the finger Dewar, samples were frozen as uncracked glasses in the flat cell by rapid freezing (see previous section) in a liquid  $N_2$  bath. 77 K actinic illumination of the optical samples was performed directly into the finger Dewar. Spectra were collected in two spectral windows (green and red wavelength regions) in order to maximize the signal-to-noise ratio in each window and to account for the different absorbance properties of PSII in each of the regions. No actinic effect was produced by the weak detecting beam of the spectrophotometer at 77 K as determined by a comparison of sequential scans of a nonilluminated sample to scans collected after 2 min of 77 K illumination with the quartz-halogen lamp; each of the sequential spectra was identical to the first nonilluminated scan and showed none of the light-induced features present in the scans collected after illumination.

**Least-Squares Fitting of Optical Difference Spectra.** The LT optical spectra were fit with a set of 2–8 Gaussian curves using a least-squares routine in Kaleidagraph version 3.0.9 graphics software. Each absorbance band was approximated by a single Gaussian line shape, and band shifts were generated from the difference between two Gaussians with equal bandwidths and amplitudes but different peak positions. All Chl absorbance amplitudes were fixed to the same value. Because of the smaller extinction coefficient of Pheo relative to Chl (20, 32, 34), Gaussian line shapes used to model Pheos were normalized by 0.6 relative to the Gaussian line shapes that represented Chls. Fits were performed on spectra plotted as a function of energy ( $\text{cm}^{-1}$ ). However, analogous fits of

spectra plotted as a function of wavelength showed that the wavelength-dependent compression of absorbance bands was insignificant over the narrow range examined (650–700 nm); therefore, all spectra are presented as a function of wavelength. To obtain fits for WT and mutant spectra simultaneously, a least-squares routine was run for each spectrum in which all of the band shifts were fixed except one which represented the chromophore that was perturbed by the mutation. To determine the positions of other chromophores, their band shifts were shifted the same amount for both spectra in an iterative manner until the best fits were obtained.

While the least-squares fits provided very good simulations of the experimental spectra, there are several factors not modeled by the fits that may account for the minor differences between the experimental and simulated spectra. The simulated absorbance bands in this study were approximated as Gaussian line shapes. This assumption ignores vibrational transitions that broaden absorbance bands toward higher energies. Although vibronic structure may have a small effect on the peak positions of the PSII chromophores, including it in the fits would have introduced an additional set of unknown parameters. In addition, in comparing WT and the D1-Q130E mutant, the bandwidth and  $\lambda_{\text{max}}$  were allowed to vary for only a single chromophore in each spectrum: the chromophore most directly affected by the mutation. This approach avoided the introduction of spectral variation lacking a known structural precedent and is supported by the assertion that Chl spectral heterogeneity is due primarily to local differences in protein environment (35, 36).

**LT EPR.** Cryogenic EPR spectra were collected on a Varian E-line EPR spectrometer equipped with an Oxford Instruments ESR 900 liquid helium cryostat. Spectrometer conditions were as follows: microwave frequency, 9.28 GHz; microwave power, 0.03 mW ( $\text{Chl}_Z^+/\text{Car}^+$  and  $\text{Y}_D^*$ ) or 20 mW ( $\text{Q}_A^-$ ); magnetic field modulation amplitude, 2 G ( $\text{Chl}_Z^+/\text{Car}^+$  and  $\text{Y}_D^*$ ) or 20 G ( $\text{Q}_A^-$ ); temperature, 30.0 K ( $\text{Chl}_Z^+/\text{Car}^+$  and  $\text{Y}_D^*$ ) or 5.9 K ( $\text{Q}_A^-$ ). The  $\text{Chl}_Z^+/\text{Car}^+$  and  $\text{Y}_D^*$  signals were collected under nonsaturating conditions. Freymy's salt was used as a standard for spin quantitation (37).

## RESULTS

To characterize the effect of a negative charge on  $\text{Q}_A^-$ , LT optical spectra of  $\text{NH}_2\text{OH}/\text{K}_3\text{Fe}(\text{CN})_6$ -treated PSII were collected before and after RT illumination. Difference spectra were obtained throughout the visible wavelength range. Results from the green and red regions are shown in Figure 1. The red wavelength region was particularly rich with absorbance signals revealing two or more band shifts in the  $\text{Q}_y$  region. Absorbance changes in the difference spectrum of the  $\text{Q}_y$  region were approximately 20–50 times smaller than the absorbance maximum of the  $\text{Q}_y$  band, which represents a superposition of all of the neutral Chls and Pheos in PSII. This ratio is reasonable considering that there are about 40 Chls and 2 Pheos per PSII core complex (30). No signals were observed in difference spectra of the Chl  $\text{Q}_x$  region, probably because they were too small relative to the noise to be detected in these experiments. However, a flat baseline in difference spectra of the  $\text{Q}_x$  region provided an indication of a good subtraction in both the  $\text{Q}_x$  region and the  $\text{Q}_y$  region. No  $\text{Q}_y$  absorbance bands were expected to be

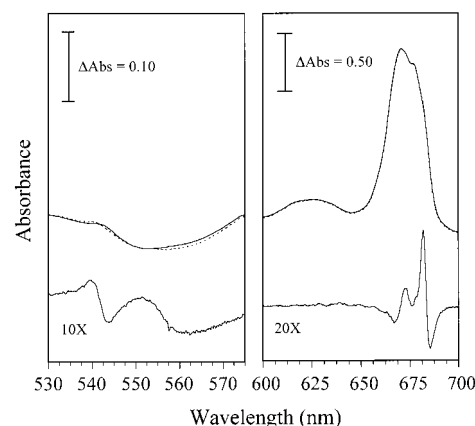


FIGURE 1: Low-temperature optical spectra of  $\text{NH}_2\text{OH}/\text{K}_3\text{Fe}(\text{CN})_6$ -treated WT PSII. The left and right panels show the green and red wavelength regions, respectively (note the different abscissa scales). The top two traces in each panel are RT illuminated (—) and nonilluminated (---) scans; the bottom traces are the illuminated minus nonilluminated difference spectra magnified 10- or 20-fold (—).

Table 1: Summary of Signal Yields in Optical and EPR Difference Spectra of PSII Core Complexes Treated with 1 Equiv of  $\text{K}_3\text{Fe}(\text{CN})_6$  and then 1000 Equiv of  $\text{NH}_2\text{OH}$

difference spectrum	EPR signal area <sup>a</sup>			optical signal height <sup>b</sup>
	$\text{Q}_A^-$	$\text{Chl}_Z^+/\text{Car}^+$	$\text{Y}_D^*/\text{Y}_Z^*$	$\text{Q}_A^-$
RT $h\nu$ minus dark	74	<3	0	90
RT $h\nu/77$ K $h\nu$ minus RT $h\nu$	<12	23	<3	10
77 K $h\nu$ minus dark	80	81	<10	79

<sup>a</sup> The  $\text{Q}_A^-$  EPR signal areas are reported as a percentage relative to the integrated area of  $\text{Q}_A^-$  after 77 K illumination and RT illumination. The  $\text{Chl}_Z^+/\text{Car}^+$  and  $\text{Y}_D^*$  EPR signal areas are relative to the double integrated area of the full yield of  $\text{Y}_D^*$  produced in a sample with matched concentration (see Materials and Methods). The values of percent change in EPR signal areas have an uncertainty of  $\pm 10$  for  $\text{Q}_A^-$  and  $\pm 5$  for  $\text{Chl}_Z^+/\text{Car}^+$  and  $\text{Y}_D^*$ . <sup>b</sup> The  $\text{Q}_A^-$  optical signal heights are reported as a percentage relative to the peak-to-trough height of the C550 signal in the 77 K illuminated and RT illuminated minus dark difference spectrum. The values of percent change in optical signal height have an uncertainty of  $\pm 10$ .

enhanced or bleached (only shifted) as a result of  $\text{Q}_A^-$  reduction, so the  $\text{Q}_A^-/\text{Q}_A$  subtraction was performed such that the integrated area under the difference spectrum in the  $\text{Q}_y$  region was zero.

The amount of  $\text{Q}_A^-$  trapped by the RT illumination protocol was quantified on the basis of the C550 signal in the illuminated minus nonilluminated difference spectrum in the green wavelength region (Table 1). This derivative-shaped feature with  $\lambda_{\text{max}} = 540$  nm and  $\lambda_{\text{min}} = 544$  nm (sharpened at 77 K relative to RT) has been assigned previously as an electrochromic shift of the Pheo absorbance band caused by a perturbation of its local electric field due to the change in redox state of  $\text{Q}_A^-$  (38, 39). In addition, the absence of any signal at 559 nm indicates that the redox state of cyt  $b_{559}$  was not changed by RT illumination. In general, the red wavelength region contained the sharpest and most well-resolved features in the  $\text{Q}_A^-$  minus  $\text{Q}_A$  spectrum. Furthermore, its spectral complexity indicated the contribution of multiple chromophores. Therefore, we have focused on the  $\text{Q}_A^-$  minus  $\text{Q}_A$  difference spectrum in the  $\text{Q}_y$  region as the subject of this paper.

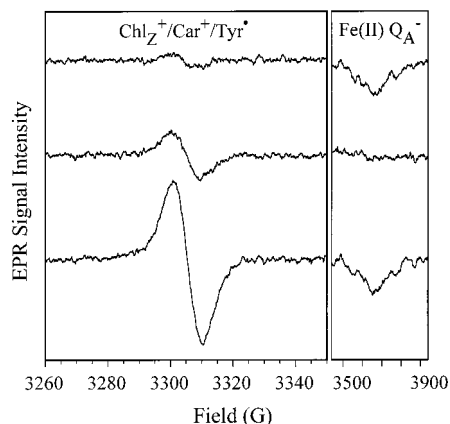


FIGURE 2: Low-temperature EPR difference spectra of  $\text{NH}_2\text{OH}/\text{K}_3\text{Fe}(\text{CN})_6$ -treated WT PSII showing the  $\text{ChlZ}^+/\text{Car}^+/\text{Tyr}^*$  (left panel) and  $\text{Fe(II)Q}_\text{A}^-$  (right panel) signals. The difference spectra are plotted as follows: RT illuminated minus nonilluminated (top trace), RT and 77 K illuminated minus RT illuminated (middle trace), and 77 K illuminated minus nonilluminated (bottom trace).

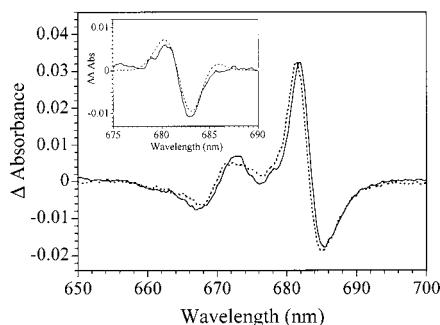


FIGURE 3: Low-temperature optical difference spectra of  $\text{NH}_2\text{OH}/\text{K}_3\text{Fe}(\text{CN})_6$ -treated WT (—) and D1-H198Q (---) PSII in the red wavelength region. The spectra are RT illuminated minus nonilluminated difference spectra. Inset: The double-difference spectrum of D1-H198Q minus WT (---) and the least-squares fit of the double-difference spectrum (—).

The yields of  $\text{Q}_\text{A}^-$  as well as  $\text{ChlZ}^+/\text{Car}^+$ ,  $\text{Y}_\text{D}^*$ , and  $\text{Y}_\text{Z}^*$  were confirmed by LT EPR spectroscopy (Figure 2, Table 1). The left panel of Figure 2 shows the  $g = 2$  region of the EPR spectrum in which oxidized Chl, Car, and tyrosine can be detected, and the right panel shows the  $g = 1.83$  region where a signal from reduced  $\text{Q}_\text{A}$  ( $\text{Fe(II)Q}_\text{A}^-$ ) can be detected (40). A significant yield of  $\text{Q}_\text{A}^-$  is present only in the first and third traces. There is essentially no contribution ( $<3\%$  of centers) from  $\text{ChlZ}^+/\text{Car}^+$ ,  $\text{Y}_\text{D}^*$ , or  $\text{Y}_\text{Z}^*$  in the  $\text{Q}_\text{A}^-$  minus  $\text{Q}_\text{A}$  spectrum (top trace). In addition, the yield of  $\text{Q}_\text{A}^-$  following RT illumination is nearly the same as that obtained by 77 K illumination alone (bottom trace). The RT/77 K illuminated minus RT illuminated spectra (middle traces) show the signals induced by 77 K illumination of the sample already illuminated at RT. These spectra reflect the number of centers that still contained oxidized  $\text{Q}_\text{A}$  (and were still capable of charge separation) after the RT illumination. Thus, the small  $\text{ChlZ}^+/\text{Car}^+$  signal and negligible  $\text{Fe(II)Q}_\text{A}^-$  signal indicate that only a small amount of  $\text{Q}_\text{A}$  ( $\sim 10\text{--}20\%$ ) remained oxidized after the RT illumination.

To assign the individual chromophores contributing to the  $\text{Q}_\text{A}^-$  minus  $\text{Q}_\text{A}$  spectrum, several site-directed mutants were compared to WT. A  $\text{Q}_\text{A}^-$  minus  $\text{Q}_\text{A}$  spectrum was obtained for D1-H198Q (Figure 3), and a direct comparison to WT was made by generating a double-difference spectrum (inset

Figure 3). The fact that there is a feature in the double-difference spectrum indicates that the D1-H198Q mutation perturbs the  $\text{Q}_\text{Y}$  band of one of the chromophores in PSII and, more importantly, that the absorbance band of this chromophore is shifted by the reduction of  $\text{Q}_\text{A}$ . The observation of a spectral shift in the  $\text{Q}_\text{Y}$  region of D1-H198Q relative to WT is consistent with previous studies on this same mutant which show a perturbation of the WT spectrum of  $\text{P}_\text{A}$  in both the Soret and  $\text{Q}_\text{Y}$  regions although the major change observed in the  $\text{Q}_\text{Y}$  region is a band shift from 672 to 669 nm (Diner, B. A., Nixon, P. J., Coleman, W. J., Schlodder, E., Rappaport, F., Lavergne, J., and Chisholm, D. A., manuscript in preparation). The difference between the  $\text{Q}_\text{A}^-$  minus  $\text{Q}_\text{A}$  spectrum of WT and D1-H198Q was confirmed by  $\text{ChlZ}^+/\text{Car}^+/\text{Q}_\text{A}^-$  minus  $\text{ChlZ}/\text{Car}/\text{Q}_\text{A}$  spectra which are also different in D1-H198Q relative to WT (data not shown). These latter spectra are not included in this paper because they contain contributions from at least three charged species and reflect electrostatic influences from cofactors on both the acceptor and donor sides of PSII. Although it appears that there is also a small shift at  $\sim 670$  nm in the  $\text{Q}_\text{A}^-$  minus  $\text{Q}_\text{A}$  spectrum of D1-H198Q as compared to WT, this represents spectral variation rather than a real mutation-induced shift as evidenced by the absence of well-defined band shifts in the double-difference spectrum.

If the D1-H198Q mutation shifts the absorbance of a chromophore in only one of the  $\text{Q}_\text{A}$  redox states (reduced or oxidized), then the double-difference spectrum would take the shape of a symmetric Gaussian first-derivative-shaped feature. On the other hand, if the D1-H198Q mutation influences a chromophore in both  $\text{Q}_\text{A}$  redox states, then the double-difference spectrum would represent a shift of a band shift—giving a feature with three components (Gaussian second-derivative-shaped feature) with the symmetry of the feature depending on the relative sizes of the mutation-induced shifts in each of the redox states of  $\text{Q}_\text{A}$ . Clearly, the D1-H198Q minus WT double-difference spectrum is asymmetric, resembling the latter of these cases more closely. (Here, we make the simplifying assumption that the mutation affects only a single chromophore.) Consequently, it was best fit as the difference of two band shifts, each having different shift magnitudes (inset of Figure 3). The parameters obtained from this fit provide an approximate bandwidth for one of the RC Chls in PSII and also define the positions of its absorbance band in both  $\text{Q}_\text{A}$  redox states for WT and D1-H198Q. From this analysis, it was determined that in WT PSII a Chl with a bandwidth  $\sim 4$  nm absorbs at 683.5 nm when  $\text{Q}_\text{A}$  is oxidized and shifts to 682.8 nm upon  $\text{Q}_\text{A}$  reduction. In the D1-H198Q mutant, these positions are 682.8 and 681.6 nm, respectively. In the above analysis, we considered all of the RC chromophores as monomeric species with no significant electronic coupling among chromophores. This approximation enabled us to eliminate the effect of exciton coupling on the chromophore positions and intensities. While the extent of chromophore coupling within the PSII RC is still currently under debate, there is ample experimental evidence supporting a weak coupling model (see ref 20 and references therein).

Additional information about the chromophores contributing to the  $\text{Q}_\text{A}^-$  minus  $\text{Q}_\text{A}$  spectrum was obtained from a comparison of the  $\text{Q}_\text{A}^-$  minus  $\text{Q}_\text{A}$  spectra for WT and the mutant, D1-Q130E (Figure 4). The glutamine at position

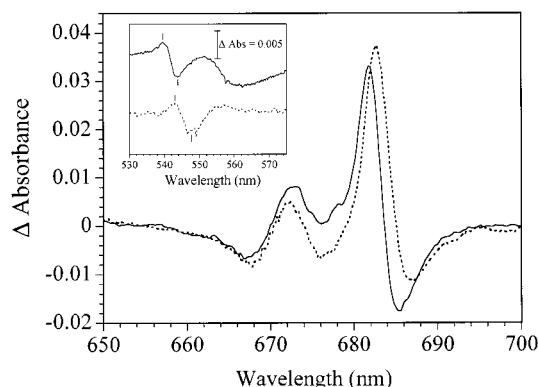


FIGURE 4: Low-temperature optical difference spectra of  $\text{NH}_2\text{OH}/\text{K}_3\text{Fe}(\text{CN})_6$ -treated WT (—) and D1-Q130E (---) PSII in the red wavelength region. The spectra are RT illuminated minus nonilluminated difference spectra. Inset: Same difference spectra in the green wavelength region.

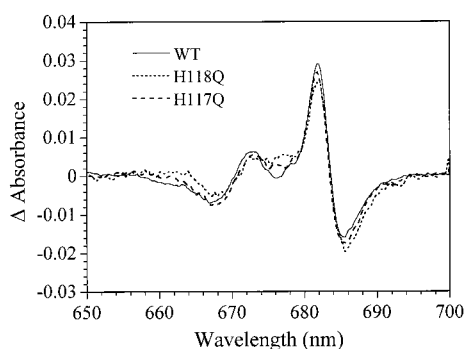


FIGURE 5: Low-temperature optical difference spectra of  $\text{NH}_2\text{OH}/\text{K}_3\text{Fe}(\text{CN})_6$ -treated WT (—), D1-H118Q (---), and D2-H117Q (— · —) PSII in the red wavelength region. The spectra are RT illuminated minus nonilluminated difference spectra.

D1-130 in WT is closely associated with the active (D1) Pheo of PSII in an interaction which is modified by replacement of the native residue with glutamic acid (28). The difference between the  $\text{Q}_\text{A}^-$  minus  $\text{Q}_\text{A}$  spectra for WT and D1-Q130E is significantly larger than that seen for WT versus D1-H198Q indicating that the  $\text{Q}_\text{y}$  absorbance band of the D1-side Pheo is shifted by  $\text{Q}_\text{A}$  reduction and also by the D1-Q130E mutation. Furthermore, the C550 signal associated with this Pheo is red-shifted by  $\sim 2.5$  nm in D1-Q130E relative to WT (inset of Figure 4), consistent with Pheo  $\text{Q}_\text{x}$  band bleaching experiments in PSII (28), optical measurements of L-branch BPheo mutants in the BRC (41), and assignment of the C550 signal as originating from Pheo (38, 39). A similar shift of the Pheo  $\text{Q}_\text{x}$  absorbance band also has been observed at RT (Force, D. A., Nixon, P. J., and Diner, B. A., unpublished results). The broad negative feature at  $\sim 562$  nm that appears in the WT spectrum (Figure 4, inset) is not a consequence of  $\text{Q}_\text{A}$  reduction; rather, it represents a baseline artifact.

The assignment of chromophores contributing to the  $\text{Q}_\text{A}^-$  minus  $\text{Q}_\text{A}$  spectrum was supported by analysis of  $\text{Q}_\text{A}^-$  minus  $\text{Q}_\text{A}$  spectra collected from two other mutants, D1-H118Q and D2-H117Q (Figure 5). D1-H118 and D2-H117 are the respective axial histidine ligands of the accessory Chls,  $\text{Chl}_\text{Z}$  and  $\text{Chl}_\text{D}$ . Both of these mutants have spectra identical to WT, a characteristic that is consistent with the interpretation that neither of the absorbance bands of  $\text{Chl}_\text{Z}$  or  $\text{Chl}_\text{D}$  is influenced by the redox state of  $\text{Q}_\text{A}$ .

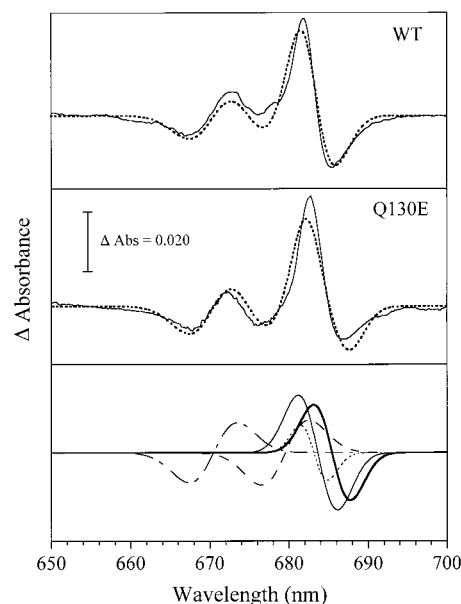


FIGURE 6: Four-chromophore fits of absorbance changes induced by  $\text{Q}_\text{A}$  reduction in WT (top panel) and D1-Q130E (middle panel) PSII. The top two panels show the experimental spectra (—) and the least-squares fits (---), and the bottom panel shows the individual band shifts that comprise the least-squares fits. The band shifts for Pheo(D2) (---),  $\text{B}_\text{A}$  (····), and Chl (— · —) are the same for WT and Q130E; the band shift for Pheo(D1) is different for WT (thin solid line) relative to D1-Q130E (bold solid line).

Because of the more complicated nature of the mutation-induced shifts in the  $\text{Q}_\text{A}^-$  minus  $\text{Q}_\text{A}$  spectrum, the WT and D1-Q130E spectra were fit simultaneously instead of fitting the double-difference spectrum. This strategy of global analysis decreases the number of independent parameters per data set (spectrum), thereby increasing the accuracy and confidence in each parameter (24). In addition, this approach allowed for the assignment of another chromophore that contributes to the  $\text{Q}_\text{A}^-$  minus  $\text{Q}_\text{A}$  spectrum besides Pheo(D1). Figure 6 shows the experimental spectrum and the best fit for WT (top panel) and D1-Q130E (middle panel) PSII; the bottom panel displays the individual band shifts which, when summed, give the spectral fits.

The procedure for assigning chromophores in PSII based on  $\text{Q}_\text{A}^-$  minus  $\text{Q}_\text{A}$  difference spectra was as follows. (i) The position and bandwidth of one Chl was determined from the least-squares fit of the H198Q minus WT  $\text{Q}_\text{A}^-$  minus  $\text{Q}_\text{A}$  double-difference spectrum (see above). (ii) The isolated band shift centered at  $\sim 670$  nm was fit as the difference of two Gaussians with a bandwidth of 7 nm and positions of 669.3 and 671.6 nm when  $\text{Q}_\text{A}$  is oxidized and reduced, respectively. The chromophore responsible for this band shift was assigned as the inactive (D2) Pheo based on a previous identification of the inactive Pheo absorbance band at  $\sim 670$  nm (22), the lack of any difference between WT and D1-Q130E in this region of the spectrum, and the close proximity of Pheo(D2) to  $\text{Q}_\text{A}$ . (iii) The Chl and Pheo(D2) positions and bandwidths were fixed, and the rest of the spectrum from 650 to 700 nm was simultaneously fit for WT and D1-Q130E to 1, 2, or 3 band shifts. During the least-squares fitting procedure, only one band shift was allowed to differ between D1-Q130E and WT corresponding to the absorbance band of Pheo(D1). After obtaining the best fits of the  $\text{Q}_\text{A}^-$  minus  $\text{Q}_\text{A}$  spectra, it was determined that fits with a total of 3 band shifts did not reproduce the experimental spectrum as well

Table 2: Summary of Absorbance Band Parameters for WT PSII Assigned from Four-Chromophore Least-Squares Fits of  $Q_A^-$  Minus  $Q_A$  Difference Spectra from WT and Mutant PSII

chromophore	$\lambda_{\max}$ (nm) ( $Q_A/Q_A^-$ )	bandwidth (fwhm, nm)
active Pheo(D1)	685.6/681.7 ( $\pm 0.2$ )	5
inactive Pheo(D2)	669.3/671.6 ( $\pm 0.3$ )	7
$B_A$ Chl	683.5 <sup>b</sup> /682.8 ( $\pm 0.1$ )	4
Chl <sup>a</sup>	678.8/680.2 ( $\pm 0.3$ )	7
Chl <sub>Z</sub>	not shifted by $Q_A^-$	
Chl <sub>D</sub>	not shifted by $Q_A^-$	
$P_A$ Chl	672 <sup>b</sup> —; not shifted by $Q_A^-$	
$P_B$ Chl	not shifted by $Q_A^-$	

<sup>a</sup> Tentatively assigned as  $B_B$  Chl. <sup>b</sup> Diner, B. A., Nixon, P. J., Coleman, W. J., Schlodder, E., Rappaport, F., Lavergne, J., and Chisholm, D. A., manuscript in preparation.

as the 4-band shift fits and that fits with 5 band shifts did not reach a local minimum, indicating that 5 band shift fits contain too many parameters to be constrained by the available data. The 4-band shift fits showed a good resemblance to the experimental spectra and could be justified based on analogy to the BRC structure (Table 2). This result indicates that besides a Chl, Pheo(D2), and Pheo(D1), an additional Chl is influenced by  $Q_A$  reduction.

The additional Chl identified from the 4-chromophore fits was assigned a bandwidth of 7 nm. This bandwidth is within the range of values determined from other LT absorbance spectra (20, 21, 42, 43) and simplified our fits by reducing the number of variables. The individual Chl absorbance maximum obtained from the 4-chromophore fit is  $\sim 0.03$  absorbance units, a value that is in agreement with the overall  $Q_y$  absorbance maximum of the PSII sample ( $\sim 1.3$ ) considering about 40 overlapping Chls per PSII complex. The spectral parameters and chromophore assignments obtained from the fits of WT and D1-Q130E are summarized in Table 2.

Of the two Pheos and two Chls comprising the  $Q_A^-$  minus  $Q_A$  fits, each chromophore type has one blue-shifting and one red-shifting band. The red-shifting chromophores are the 7 nm bandwidth Chl and Pheo(D2), and the blue-shifting chromophores are the Chl identified from the WT/D1-H198Q comparison and Pheo(D1). The dominant band shift in both WT and D1-Q130E spectra is the chromophore affected by the D1-Q130E mutation, Pheo(D1). When  $Q_A$  is oxidized, the Pheo(D1) absorbance band is at 685.6 nm (WT) or 686.9 nm (D1-Q130E), and when  $Q_A$  is reduced the Pheo(D1) band is blue-shifted by 3.9 nm in WT and 2.9 nm in D1-Q130E. Thus, the effect of the D1-Q130E mutation on the  $Q_A^-$  minus  $Q_A$  spectrum is a red-shift of the Pheo(D1) absorbance band and a slight decrease in the intensity of the Pheo(D1) band shift.

## DISCUSSION

The spectra presented in this paper combine the advantages of LT measurement of well-defined charge-separated states and site-directed mutation to provide detailed information about individual chromophores in PSII. The use of intact PSII core complexes purified by mild detergent extraction (30) in this study ensured minimal perturbation of the RC and its chromophores during the isolation procedure. Chemical treatment and 77 K measurement enabled us to trap a well-defined redox state change for the bound quinone,  $Q_A$ .

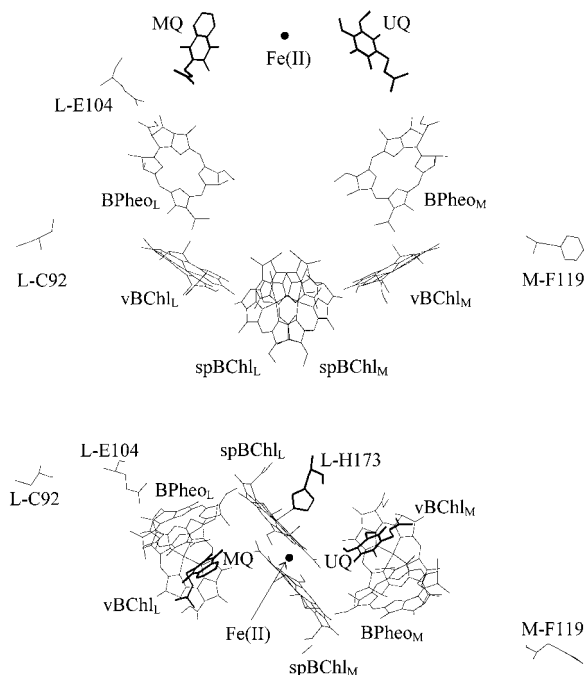


FIGURE 7: Top: Side-on view (in the membrane plane). Bottom: Stromal-side view (along the membrane normal) of the chromophores and relevant amino acids in the BRC from *R. viridis* (44). The chromophores are indicated as follows: L and M branch special pair BChls (spBChl<sub>L</sub>, spBChl<sub>M</sub>), L and M branch monomeric "voyeur" BChls (vBChl<sub>L</sub>, vBChl<sub>M</sub>), L and M branch BPheos (BPheo<sub>L</sub>, BPheo<sub>M</sub>), ubiquinone (UQ, analogous to  $Q_B$ ), menaquinone (MQ, analogous to  $Q_A$ ), and the non-heme Fe(II) atom (Fe(II)). The amino acids are L-C92 (the homologue of D1-H118, the axial ligand to Chl<sub>Z</sub>), M-F119 (the homologue of D2-H117, the axial ligand to Chl<sub>D</sub>), L-E104 (the homologue of D1-Q130E), and L-H173 (the homologue of D1-H198).

In addition, individual chromophore absorbance bands are narrower at lower temperatures (24), making them more readily resolved. Previous studies of the  $P680^+-Q_A^-$  state have shown that spectral resolution of 77 K absorbance spectra is significantly improved over RT spectra as a consequence of this spectral narrowing (21).

**Relevance of the Bacterial Reaction Center Structure.** In evaluating the least-squares fits of the  $Q_A^-$  minus  $Q_A$  difference spectra and assigning deconvoluted absorbance bands to specific chromophores, it is extremely useful to take advantage of the known structural similarity between the BRC and PSII. We used the crystal structure of the BRC from *R. viridis* (44) (PDB 1PRC) as a template for assessing the likelihood of electrochromic interactions in PSII. Two views of the BRC chromophores including the special pair BChl dimer, the voyeur accessory BChls, the BPheos, and the  $Q_A$  and  $Q_B$  homologues (menaquinone (MQ) and ubiquinone, respectively) are presented in Figure 7. In both views, three BRC amino acids are highlighted because of the relevance of their PSII homologues: L-C92 and M-F119 are homologous to D1-H118 and D2-H117, respectively (14), and L-E104 is homologous to D1-Q130. The stromal side view also shows L-H173, the homologue of D1-H198. To aid in the comparison of charge–chromophore interactions, the center-to-center distances between MQ and each of the BRC chromophores and the angles between MQ and the y- and x-axes of the BRC chromophores were estimated from the BRC coordinates; the values are summarized in Table 3.

Table 3: Summary of Distances (Å) and Orientations of Menaquinone (MQ) Relative to Each of the Chromophores in the BRC Measured from the 2.30-Å Crystal Structure of *R. viridis* (PDB 1PRC)<sup>a</sup>

chromophore	distance from MQ (Q <sub>A</sub> ) <sup>b</sup>	orientation with respect to	orientation with respect to	predicted shift of Q <sub>y</sub> by Q <sub>A</sub> <sup>-i</sup>
		y-axis <sup>g</sup> (deg)	x-axis <sup>g</sup> (deg)	
special pair BChl (L) <sup>c</sup>	28.0	-74	+27	blue
special pair BChl (M) <sup>c</sup>	28.3	-56	+34	blue
voyeur BChl (L) <sup>c</sup>	23.8	-72	+68	blue
voyeur BChl (M) <sup>c</sup>	30.0	+79	-88	red
BPheo (L) <sup>d</sup>	14.4	+47	+82	red
BPheo (M) <sup>d</sup>	23.3	+60	-55	red
L-C92 [Chl <sub>z</sub> ] <sup>e</sup>	32.8	NA	NA	NA
M-F119 [Chl <sub>D</sub> ] <sup>f</sup>	42.0	NA	NA	NA

<sup>a</sup> For comparison to PSII: MQ corresponds to Q<sub>A</sub>, L-C92 is homologous to D1-H118, M-F119 is homologous to D2-H117, L-E104 is homologous to D1-Q130E, and L-H173 is homologous to D1-H198.

<sup>b</sup> Distances were calculated from the center of the MQ ring by averaging the distances to three nonadjacent carbons in the ring. <sup>c</sup> Distances were measured from the magnesium in the center of the BChl rings.

<sup>d</sup> Distances were estimated to the center of the BPheo macrocycles by averaging the distances to the four pyrrole nitrogens. <sup>e</sup> Distances were estimated to the center of the Chl<sub>z</sub> macrocycle by using the sulfur atom of L-C92, which is expected to be in a comparable position to the nitrogen of D1-H118 that serves as the axial ligand to Chl<sub>z</sub>.

<sup>f</sup> Distances were estimated to the center of the Chl<sub>D</sub> macrocycle by using the average of the distances to the C3 and C5 positions of the phenylalanine ring of M-F119. These carbon atoms, four atoms away from the protein backbone, are expected to be in comparable positions to the nitrogen of D2-H117, which serves as the axial ligand to Chl<sub>D</sub>. <sup>g</sup> Angles were measured between the center of MQ ring<sup>b</sup> and the y-axis or x-axis of each chromophore<sup>h</sup> using the center of the BChl or BPheo ring as the vertex of the angle. Orientations of the y-axis in which Q<sub>A</sub> was closer to ring III and orientations of the x-axis in which Q<sub>A</sub> was closer to ring IV were arbitrarily assigned as positive angles. <sup>h</sup> The y-axis was approximated as a vector between the pyrrole nitrogens of rings I and III, and the x-axis was approximated as a vector between the pyrrole nitrogens of rings II and IV. <sup>i</sup> The predicted Q<sub>y</sub> band shift directions were determined using the measured orientation of MQ relative to the y-axis of each chromophore as described in ref 19. A positive angle indicates that MQ is closer to ring III, and a negative angle indicates that MQ is closer to ring I.

**Chromophore Assignments.** The comparison of Q<sub>A</sub><sup>-</sup> minus Q<sub>A</sub> spectra lacking contributions from oxidized electron donors for WT and several site-directed mutants of PSII makes possible a straightforward identification of chromophores that are influenced by the redox state of Q<sub>A</sub>. Specifically, at least one Chl and one Pheo are influenced by the negative charge on Q<sub>A</sub> as evidenced by mutation-induced shifts of their absorbance bands in Q<sub>A</sub><sup>-</sup> minus Q<sub>A</sub> spectra of D1-H198Q and D1-Q130E. In the case of the Chl, the shift induced by the D1-H198Q mutation is a subtle change from the WT Q<sub>A</sub><sup>-</sup> minus Q<sub>A</sub> spectrum, but the perturbation is readily apparent in the double-difference spectrum. The double-difference spectrum is an isolated representation of the effect of the mutation on the response of this Chl to the redox state of Q<sub>A</sub>; as such, this spectrum contains information about the position and shape of the Chl absorbance band that can be extracted from least-squares fits of the spectrum.

Because the D1-H198Q mutation replaces a histidine that is homologous to the axial ligand of the L-branch special pair BChl, the expected identity of the Chl influenced by this mutation is the D1-side monomeric core Chl, P<sub>A</sub>. Recent LT (5 and 120 K) optical experiments examining P680<sup>+</sup>

minus P680 spectra and <sup>3</sup>P680 minus P680 spectra for WT and D1-H198Q have supported this expectation by revealing a ~3 nm blue shift of the P<sub>A</sub> absorbance band in D1-H198Q (669 nm) relative to WT (672 nm) (Diner, B. A., Nixon, P. J., Coleman, W. J., Schlodder, E., Rappaport, F., Lavergne, J., and Chisholm, D. A., manuscript in preparation). However, the Q<sub>A</sub><sup>-</sup> minus Q<sub>A</sub> spectra presented here show a very small shift associated with the D1-H198Q mutation (~0.7 nm). In addition, the λ<sub>max</sub> of the Chl that is affected by both D1-H198Q mutation and Q<sub>A</sub> reduction is 683.5 nm rather than 672 nm. These observations, along with the large distance expected between Q<sub>A</sub> and P<sub>A</sub> based on BRC homology (28.0 Å, Table 3), argue against the assignment of the 683.5 nm Chl as P<sub>A</sub>. Interestingly, the aforementioned studies of Diner et al. have identified a second Chl associated with the absorbance spectrum of <sup>3</sup>P680 and P<sub>A</sub> that absorbs at 682–684 nm and is assigned to the D1-side monomeric accessory Chl, B<sub>A</sub>. Because of the closer proximity of the L-branch voyeur BChl to MQ, the previous assignment of B<sub>A</sub> at 682–684 nm, and the observation that B<sub>A</sub> is associated with P<sub>A</sub>, we assign B<sub>A</sub> as the ~4-nm-wide absorbance band that absorbs at 683.5 nm when Q<sub>A</sub> is oxidized.

The assignment of B<sub>A</sub> based on changes induced by the D1-H198Q mutation requires that there is a secondary effect of this mutation on the B<sub>A</sub> Chl in addition to the effect on the P<sub>A</sub> Chl which is directly ligated by D1-H198. There are at least two possibilities that can explain this observation. One explanation is that the P680 Chls are excitonically coupled such that there is overlap between the electronic transitions of P<sub>A</sub> and B<sub>A</sub>. If this is the case, then the absorbance profile of both chromophores would be expected to be perturbed by the D1-H198Q mutation, although the extent to which B<sub>A</sub> is affected would depend on the strength of the exciton coupling. Alternatively, the secondary effect of the mutation on B<sub>A</sub> may be a consequence of structural perturbation of the protein that affects not only P<sub>A</sub> but also B<sub>A</sub>, leading to a large shift of the P<sub>A</sub> absorbance band and a small shift of the B<sub>A</sub> absorbance band.

The fact that the D1-Q130E mutation changes the Q<sub>y</sub> band in the Q<sub>A</sub><sup>-</sup> minus Q<sub>A</sub> spectrum relative to WT confirms that Pheo(D1) is affected by the reduction of Q<sub>A</sub>. Giorgi et al. showed that the D1-Q130E mutation affects the Pheo(D1) Q<sub>x</sub> absorbance band via a 2.5-nm red shift (28), and this Pheo is expected to be very close to Q<sub>A</sub> (Table 3) so the interaction between Q<sub>A</sub><sup>-</sup> and Pheo(D1) is not surprising. By utilizing simultaneous least-squares fits of WT and D1-Q130E Q<sub>A</sub><sup>-</sup> minus Q<sub>A</sub> spectra and including the parameters obtained for B<sub>A</sub> and Pheo(D2), we were able to identify and assign the absorbance bands of Pheo(D1) and another Chl associated with the PSII RC (Table 2). Our spectral analysis shows that the mutation-induced red shift of the Pheo(D1) Q<sub>x</sub> band is paralleled by a red shift of the Q<sub>y</sub> band. The difference between the Pheo(D1) absorbance band in WT versus D1-Q130E may represent a subtle change in the electronic environment around Pheo(D1) caused by the mutation. The Q to E mutation probably changes the hydrogen-bonding interaction between the D1-130 amino acid and Pheo(D1). Both of these factors would be expected to influence the interaction of Pheo(D1) with Q<sub>A</sub> in its oxidized and reduced forms. This phenomenon reflects an interesting electronic and/or structural modification of PSII that is relevant to the

efficiency of charge separation (28) and requires further study.

The fourth chromophore that is perturbed by a negative charge on  $Q_A$  is a red-shifting Chl absorbing at 678.8 nm. The possible assignments for this Chl are the remaining Chls in the PSII RC core [ $P_A$ ,  $P_B$ ,  $B_B$ ,  $Chl_Z$ , or  $Chl_D$ ] or an additional Chl at the perimeter of the D1/D2 complex in close enough proximity to  $Q_A$  to be influenced by  $Q_A$  reduction. The identity of the 678.8-nm Chl is not likely to be a Chl in CP47 based on the positions of the CP47 Chl electron densities assigned in the 8-Å structure of the CP47 RC (12). The assignment of a CP43 Chl cannot be ruled out from the CP47 RC structure, but a CP43 orientation symmetric to CP47 would also put the closest CP43 Chl a significant distance away from  $Q_A$ . Assuming homology to the BRC and based on distance alone,  $Chl_Z$  and  $Chl_D$  are the least likely of the RC Chls to be affected by  $Q_A^-$ : the homologues of the amino acids that ligate  $Chl_Z/Chl_D$  are 32.8/42.0 Å from MQ (Table 3). Furthermore, mutants in which the  $Chl_Z/Chl_D$  axial ligands are perturbed (D1-H118Q/D2-H117Q) do not show a difference in their  $Q_A^-$  minus  $Q_A$  spectra relative to WT (Figure 5). Similarly, the absence of a shifting band at 672 nm in the D1-H198Q  $Q_A^-$  minus  $Q_A$  spectrum rules out  $P_A$  as a candidate. Because  $P_B$  is expected to be in close proximity to  $P_A$  and to have a similar distance with respect to  $Q_A$ , it seems unlikely that  $P_B$  is the fourth chromophore affected by  $Q_A$  reduction. In the BRC, the remaining RC BChl, the M-branch voyeur BChl, is slightly farther from MQ than the special pair BChls; thus,  $B_B$  does not appear to be ideally situated for interaction with  $Q_A^-$ . On the other hand, a distance between  $B_B$  (and  $B_A$ ) and  $Q_A$  predicted based on the BRC structure may be flawed given that the axial ligands of the voyeur BChls are not conserved in PSII (9). A difference in the positions of the voyeur BChls and  $B_A/B_B$  relative to the quinones (23) could result in a stronger  $B_B-Q_A^-$  interaction in PSII. One factor favoring the assignment of  $B_B$  as the fourth chromophore shifted by  $Q_A$  reduction is the fact that it supports a model of the RC in which symmetry plays a significant role: the four chromophores shifted by  $Q_A$  reduction would be both Pheos,  $B_A$ , and  $B_B$ , with each pair related by 2-fold symmetry. We tentatively assign  $B_B$  as the second Chl shifted by  $Q_A$  reduction.

In general, the chromophore assignments are consistent with a structure of the PSII RC that is similar to that of the BRC. For example, the dominant band shift contributing to the  $Q_A^-$  minus  $Q_A$  spectrum corresponds to Pheo(D1), the chromophore which is expected to be the closest to  $Q_A$  and the most strongly influenced by  $Q_A$  reduction based on the BRC structure (Table 3). Similarly, the band shift of Pheo(D2) has a significant contribution to the  $Q_A^-$  minus  $Q_A$  spectrum, and the homologous BPheo is the second closest chromophore to MQ in the BRC.  $B_A$  is also influenced by  $Q_A$  reduction, and its homolog, the L-branch voyeur BChl, is the third closest chromophore to MQ. The absence of an influence of  $Q_A^-$  on  $P_A$  revealed using the D1-H198Q mutant is consistent with a position relative to  $Q_A$  that is similar to that found between the L-branch special pair Chl and MQ: the large distance relative to MQ predicts a weak electrochromic interaction between  $P_A$  and  $Q_A$ . Also, the band shift direction observed for three of the four PSII chromophores

agrees with that predicted based on the orientation of BRC chromophores (Table 3).

Despite the consistencies between the BRC structure and the observed band shifts in PSII, there are some differences that indicate variation between the different types of reaction centers. A comparison of  $MQ^-$  minus MQ spectra for the BRC (45, 46) with the  $Q_A^-$  minus  $Q_A$  spectrum reveals that the dominant band shifts in the two spectra have opposite directionality: the main band shift in PSII is a blue shift of Pheo(D1) while the dominant band shift in the BRC is a red shift of one or both of the BPheos. Interestingly, the  $Q_x$  region band shift corresponding to Pheo(D1) is a blue shift, but the  $Q_x$  band shift for Pheo(D2) is a red shift (Force, D. A., and Diner, B. A., unpublished results). Thus, the shift directionalities in the  $Q_x$  and  $Q_y$  regions are the same for a given Pheo within PSII. Another difference between PSII and the BRC is that, in the  $MQ^-$  minus MQ spectrum, band shifts from the special pair BChls contribute more to the spectrum than band shifts of the voyeur BChls (45, 46). This is in contrast to our assignments for PSII in which  $B_A$  and  $B_B$  contribute to the  $Q_A^-$  minus  $Q_A$  spectrum and  $P_A$  and  $P_B$  do not contribute significantly.

There are several factors that may explain the spectral differences between PSII and the BRC. One possibility is that the orientations of the chromophores diverge between the BRC and PSII, an interpretation that has been applied before to the Pheos and  $Q_A$  in PSII (20, 23). Such variations would change the orientation of chromophore  $Q_y$  transition dipoles with respect to  $Q_A^-$ , giving rise to band shifts with different intensities or directionality relative to the BRC. This influence may contribute to the different responses of  $P_A$  and the L-branch special pair BChl to  $Q_A/MQ$  reduction because  $P_A$  and  $P_B$  are known to have a different position than the special pair BChls (12), and a different orientation is likely (12, 23). Another possibility is that the  $Q_y$  transition dipoles for the PSII Chls are different from those of the BRC BChls (19). If this is true, then the distribution of electron density within the chlorin rings in the ground and excited states would be different in PSII versus the BRC and would lead to different sensitivities of the chromophores to nearby charged species. In addition, local differences in the dielectric constant of the protein medium are likely to contribute because changes in dielectric strength modulate the magnitude of electrochromic band shifts. In fact, it has been shown that the active and inactive BPheo branches in the BRC display significant asymmetry: the effective dielectric constant along the L-branch of the BRC is significantly higher than that of the M-branch, resulting in smaller electrochromic shifts of the L-branch chromophores (47). Combined with a different position for  $B_B$  than the M-branch voyeur BChl (see above), a low dielectric between  $B_B$  and  $Q_A$  could explain the significant shift of the  $B_B Q_y$  absorbance band upon  $Q_A$  reduction.

While  $Q_A^-$  minus  $Q_A$  spectra for WT and D1-Q130E were fit and explained best using a blue-shifting Pheo(D1) as described in detail above, the spectra could be fit reasonably well by using a red-shifting Pheo(D1) model. However, this latter model is not favored because it requires a large difference between the band shift magnitudes for Pheo(D1) in WT versus D1-Q130E. Such variation is difficult to justify given the subtle nature of the D1-Q130E mutation and the fact that WT and D1-Q130E show  $Q_x$  band shifts that are

very similar in magnitude. In addition, the red-shifting Pheo-(D1) model required a weak interaction between Pheo(D1) and  $Q_A^-$ , an unlikely scenario given their expected close proximity.

**Comparison to Previous Assignments.** In the presence of neutral  $Q_A$ , the two Pheos have absorbance maxima at 669.3 and 685.6 nm. These positions are consistent with previous assignments. The photochemically inactive Pheo molecule was estimated to absorb at  $671 \pm 1$  nm (22), and other groups have suggested that one of the Pheos in PSII absorbs at about 670 nm (24, 25, 48, 49). The photochemically active Pheo has been assigned to be in the range of wavelengths between 680 and 685 nm (2, 20, 22, 23, 50–52), so our assignment is at the upper end of this range.

Despite the general agreement between our assignments and others, there are several factors that may explain why some previous studies have differed. One issue is the impact of sample preparation on spectral assignment. In fact, this has been presented as an explanation for the wide range of absorbance band assignments obtained from D1/D2/cyt  $b_{559}$  preparations (20). The detergent treatments used to prepare these stripped-down complexes by removing antenna proteins may have a significant impact on the absorbance bands of the core chromophores by perturbing the chromophore environment within the protein. There is evidence of a 4–5-nm blue shift of P680 upon conversion from intact PSII to isolated D1/D2/cyt  $b_{559}$  complexes, and it has been suggested that the low-energy ( $\sim 684$  nm) bands contributing to D1/D2/cyt  $b_{559}$  spectra may actually represent the residual fraction of centers that remain as intact PSII (20). Similarly, the  $Q_y$  band of P680 may be perturbed by the harsh chemical treatment used to produce doubly reduced  $Q_A$  (20). These arguments reinforce the importance of considering the individual PSII preparation and treatment used for chromophore assignments. The PSII core complexes used in this work consist of the RC complex plus additional associated proteins. In these preparations, the CP43 and CP47 proteins are likely to have a role in stabilizing the RC and preserving the native protein environment of RC core chromophores.

In some previous assignments of electrochromic spectra, there have been simplifying assumptions about the number and  $\lambda_{\max}$  of chromophores contributing to various spectra. For example, in one study of electrochromic band shifts of PSII it was assumed that a total of four chromophores are influenced by redox changes on  $Q_A$ ,  $Y_Z$ , and the Mn cluster (23). While these assumptions rendered the spectral fits tractable, they de-emphasized the expected symmetry of the D1 and D2 Pheos and are not consistent with our finding that both Pheos are influenced by a redox change at  $Q_A$ . Similarly, a  $Q_A^-$  minus  $Q_A$  spectrum obtained by Velthuis which displays remarkable similarity to the WT  $Q_A^-$  minus  $Q_A$  spectrum presented here was simulated with only one Chl and one Pheo absorbance band (53). In light of the observations of this paper, the assignment of only two chromophores affected by  $Q_A$  reduction is certainly an underestimation. In other studies, some chromophore positions have been restricted to ranges of previously determined values (24). Again, this approach decreases the number of variables in fits but limits the independence of assignments from previous work. In this study, we have avoided assumptions about the contributing chromophores prior to the identification of absorbance bands from spectral fits.

Some groups have used the directionality of  $Q_A^-$ -induced electrochromic shifts to argue in favor of or against similar placement of the Pheos and BPheos relative to  $Q_A$  in PSII and the BRC (23, 54). However, most descriptions of  $Q_A^-$  minus  $Q_A$  difference spectra report only one or two electrochromic shifts (20, 21, 23, 38, 39, 45, 46, 54). The data presented here show that  $Q_A$  reduction induces both blue and red shifts, and it is clear that a charge on  $Q_A$  alters the absorbance bands of more than one chromophore in PSII. Therefore, it is important to identify all of the affected chromophores and to consider the known chromophore stoichiometry differences between the complexes when evaluating the structural implications of spectral band shifts.

The identification of  $B_A$  at 683.5 nm is an assignment that has little basis for comparison because most efforts to identify RC chromophore bands have focused on the P680 Chls, the Pheos and the accessory Chls ligated by D1-H118 and D2-H117. However, efforts to assign low energy bands (682–684 nm) have been quite common, so several comments are warranted. Our assignment of  $B_A$  is in reasonable agreement with the work of Konermann et al., which simulated the temperature dependence of the RC  $Q_y$  spectrum and proposed an accessory Chl at 682.3 nm at 10 K (24, 55). On the other hand, the  $B_A$  assignment is in contrast to studies of detergent-treated PSII, which suggest that a 684 nm band corresponds to a labile Chl at the exterior of the RC (26) or  $Chl_Z$  (56), and it differs from studies which suggest that long wavelength ( $\sim 683$  nm) Chls in some D1/D2/cyt  $b_{559}$  preparations represent contaminating antenna Chls that were not completely removed during the PSII isolation procedure (24). While our results do not exclude the possibility that more than one Chl in the RC absorbs at  $\sim 684$  nm, they do show that if only one Chl in the RC absorbs at  $\sim 684$  nm it can be assigned to  $B_A$ .

The absorbance bands and cofactor interactions assigned in this paper have implications for previous interpretations of PSII structure and chromophore orientations. In general, this work supports a structural model of PSII that is similar to the BRC. In particular, it appears that there are voyeur-like accessory Chls in PSII that are not  $Chl_Z$  or  $Chl_D$ , an assessment that is consistent with the consensus that there are six rather than four Chls per D1/D2/cyt  $b_{559}$  complex (3–6, 56). This work, therefore, strengthens the assignment of monomeric accessory Chls in the 8-Å structure of PSII that correspond to the monomeric BChls of the BRC, although their positions and/or orientations in PSII may be slightly different from those in the BRC. Also, the observation that  $Chl_Z$  and  $Chl_D$  are not influenced by the redox state of  $Q_A$  supports the model of cyclic electron transfer in PSII in which the electron acceptor ( $Q_A$ ) and donor ( $Chl_Z$ ) sides are separated by more than one electron-transfer step and are bridged by the redox-active mediator, cyt  $b_{559}$  (57–59). The identification of an effect on both Pheos,  $B_A$ , and another Chl in PSII due to the charge on  $Q_A$  demonstrates the sensitivity of the PSII absorbance profile to redox changes and electron-transfer reactions within the RC.

## FUTURE PERSPECTIVES

The survey of redox-induced spectral features presented here demonstrates that there is a significant amount of structural and spectroscopic information to be obtained by

the application of optical difference spectroscopy to site-directed mutants of PSII. Despite extensive study, none of the Q<sub>y</sub> absorbance bands of Chls and Pheos in PSII had been unambiguously assigned, and LT optical spectroscopy of various redox states provides a means of assigning some of these components. The application of this technique to the many existing site-directed mutants of PSII—especially mutants in which the binding environments of RC chromophores are perturbed—will enable the assignment of all of the chromophores and provide important insights into the interactions between the cofactors in PSII.

The case presented in this paper is limited to providing information about chromophores whose absorbance bands are influenced by the redox state of Q<sub>A</sub>, but the approach utilized is shown to be generally very useful for assigning individual absorbance bands that are part of a complex group of overlapping bands. This strategy will certainly be productive in the future to examine difference spectra of other redox states such as Y<sub>Z</sub><sup>•</sup> minus Y<sub>Z</sub>, Y<sub>D</sub><sup>•</sup> minus Y<sub>D</sub>, Q<sub>B</sub><sup>•</sup> minus Q<sub>B</sub>, Chl<sub>Z</sub><sup>+</sup> minus Chl<sub>Z</sub>, and Car<sup>+</sup> minus Car in conjunction with site-directed mutants and our ever-increasing structural insight to elucidate the absorbance profile of PSII.

## ACKNOWLEDGMENT

We thank Dr. John Hartwig for the loan of the Lambda 6 UV–Vis spectrophotometer. D.H.S. thanks Dr. Veronika Szalai for many useful discussions and for critical reading of the manuscript.

## REFERENCES

- Tang, X.-S., and Satoh, K. (1985) *FEBS Lett.* 179, 60–64.
- Nanba, O., and Satoh, K. (1987) *Proc. Natl. Acad. Sci. U.S.A.* 84, 109–112.
- Gounaris, K., Chapman, D. J., Booth, P. J., Crystall, B., Giorgi, L. B., Klug, D. R., Porter, G., and Barber, J. (1990) *FEBS Lett.* 265, 88–92.
- Kobayashi, M., Maeda, H., Watanabe, T., Nakane, H., and Satoh, K. (1990) *FEBS Lett.* 260, 138–140.
- Eijkelhoff, C., Van Roon, H., Groot, M.-L., Van Grondelle, R., and Dekker, J. P. (1996) *Biochemistry* 35, 12864–12872.
- Tomo, T., Mimuro, M., Iwaki, M., Kobayashi, M., Itoh, S., and Satoh, K. (1997) *Biochim. Biophys. Acta* 1321, 21–30.
- Hankamer, B., Barber, J., and Boekema, E. J. (1997) *Annu. Rev. Plant Physiol. Plant Mol. Biol.* 48, 641–671.
- Debus, R. J. (1992) *Biochim. Biophys. Acta* 1102, 269–352.
- Michel, H., and Deisenhofer, J. (1988) *Biochemistry* 27, 1–7.
- Ruffle, S. V., Donnelly, D., Blundell, T. L., and Nugent, J. H. A. (1992) *Photosynth. Res.* 34, 287–300.
- Svensson, B., Etchebest, C., Tuffery, P., van Kan, P., Smith, J., and Styring, S. (1996) *Biochemistry* 35, 14486–14502.
- Rhee, K.-H., Morris, E. P., Barber, J., and Kühlbrandt, W. (1998) *Nature* 396, 283–286.
- Kouloulgiotis, D., Tang, X.-S., Diner, B. A., and Brudvig, G. W. (1995) *Biochemistry* 34, 2850–2856.
- Stewart, D. H., Cua, A., Chisholm, D. A., Diner, B. A., Bocian, D. F., and Brudvig, G. W. (1998) *Biochemistry* 37, 10040–10046.
- Diner, B. A., Nixon, P. J., and Farchaus, J. W. (1991) *Curr. Opin. Struct. Biol.* 1, 546–554.
- El-Kabbani, O., Chang, C.-H., Tiede, D., Norris, J., and Schiffer, M. (1991) *Biochemistry* 30, 5361–5369.
- Diner, B. A., and Babcock, G. T. (1996) in *Advances in Photosynthesis* (Ort, D. R., and Yocum, C. F., Eds.) pp 213–247, Kluwer Academic Publishers, Dordrecht.
- Tommos, C., Hoganson, C. W., Di Valentin, M., Lydakis-Simantiris, N., Dorlet, P., Westphal, K., Chu, H.-A., McCracken, J., and Babcock, G. T. (1998) *Curr. Opin. Chem. Biol.* 2, 244–252.
- Hanson, L. K., Fajer, J., Thompson, M. A., and Zerner, M. C. (1987) *J. Am. Chem. Soc.* 109, 4728–4730.
- Hillmann, B., Brettel, K., van Mieghem, F., Kamlowski, A., Rutherford, A. W., and Schlodder, E. (1995) *Biochemistry* 34, 4814–4827.
- Hillmann, B., and Schlodder, E. (1995) *Biochim. Biophys. Acta* 1231, 76–88.
- Mimuro, M., Tomo, T., Nishimura, Y., Yamazaki, I., and Satoh, K. (1995) *Biochim. Biophys. Acta* 1232, 81–88.
- Mulki, A. Y., Cherepanov, D. A., Haumann, M., and Junge, W. (1996) *Biochemistry* 35, 3093–3107.
- Konermann, L., and Holzwarth, A. R. (1996) *Biochemistry* 35, 829–842.
- Braun, P., Greenberg, B. M., and Scherz, A. (1990) *Biochemistry* 29, 10376–10387.
- Chang, H.-C., Jankowiak, R., Reddy, N. R. S., Yocum, C. F., Picorel, R., Seibert, M., and Small, G. J. (1994) *J. Phys. Chem.* 98, 7725–7735.
- Williams, J. G. K. (1988) *Methods Enzymol.* 167, 766–778.
- Giorgi, L. B., Nixon, P. J., Merry, S. A. P., Joseph, D. M., Durrant, J. R., De Las Rivas, J., Barber, J., Porter, G., and Klug, D. R. (1996) *J. Biol. Chem.* 271, 2093–2101.
- Nixon, P. J., Trost, J. T., and Diner, B. A. (1992) *Biochemistry* 31, 10859–10871.
- Tang, X.-S., and Diner, B. A. (1994) *Biochemistry* 33, 4594–4603.
- Tamura, N., and Cheniae, G. (1987) *Biochim. Biophys. Acta* 890, 179–194.
- Lichtenthaler, H. K. (1987) *Methods Enzymol.* 148, 350–382.
- Metz, J. G., Nixon, P. J., Rögner, M., Brudvig, G. W., and Diner, B. A. (1989) *Biochemistry* 28, 6960–6969.
- Shipman, L. L. (1982) in *Photosynthesis* (Govindjee, Ed.) pp 275–291, Academic Press, New York.
- Giuffra, E., Zucchelli, G., Sandonà, D., Croce, R., Cugini, D., Garlaschi, F. M., Bassi, R., and Jennings, R. C. (1997) *Biochemistry* 36, 12984–12993.
- Simonetto, R., Crimi, M., Sandonà, D., Croce, R., Cinque, G., Breton, J., and Bassi, R. (1999) *Biochemistry* 38, 12974–12983.
- Babcock, G. T., Ghanotakis, D. F., Ke, B., and Diner, B. A. (1983) *Biochim. Biophys. Acta* 723, 276–286.
- van Gorkom, H. J., Tamm, J. J., and Haveman, J. (1974) *Biochim. Biophys. Acta* 347, 417–438.
- van Gorkom, H. J. (1974) *Biochim. Biophys. Acta* 347, 439–442.
- Miller, A.-F., and Brudvig, G. W. (1991) *Biochim. Biophys. Acta* 1056, 1–18.
- Bylina, E. J., Kirmaier, C., McDowell, L., Holten, D., and Youvan, D. C. (1988) *Nature* 336, 182–184.
- van Kan, P. J. M., Otte, S. C. M., Kleinharenbrink, A. M., Nieveen, M. C., Aartsma, T. J., and van Gorkom, H. J. (1990) *Biochim. Biophys. Acta* 1020, 146–152.
- Zucchelli, G., Garlaschi, F. M., and Jennings, R. C. (1996) *Biochemistry* 35, 16247–16254.
- Deisenhofer, J., Epp, O., Miki, K., Huber, R., and Michel, H. (1985) *Nature* 318, 618–624.
- Vermeglio, A., and Clayton, R. K. (1977) *Biochim. Biophys. Acta* 461, 159–165.
- Shopes, R. J., and Wraight, C. A. (1985) *Biochim. Biophys. Acta* 806, 348–356.
- Steffen, M. A., Lao, K., and Boxer, S. G. (1994) *Science* 264, 810–816.
- Klimov, V. V., Klevanik, A. V., Shuvalov, V. A., and Krasnovsky, A. A. (1977) *FEBS Lett.* 82, 183–186.
- Shuvalov, V. A., Heber, U., and Schreiber, U. (1989) *FEBS Lett.* 258, 27–31.
- Ganago, I. B., Klimov, V. V., Ganago, A. O., Shuvalov, V. A., and Erokhin, Y. E. (1982) *FEBS Lett.* 140, 127–130.
- Tang, D., Jankowiak, R., Seibert, M., Yocum, C. F., and Small, G. J. (1990) *J. Phys. Chem.* 94, 6519–6522.
- van der Vos, R., van Leewen, P. J., Braun, P., and Hoff, A. J. (1992) *Biochim. Biophys. Acta* 1140, 184–198.

53. Velthuys, B. R. (1988) *Biochim. Biophys. Acta* 933, 249–257.
54. Cherepanov, D. A., Haumann, M., Junge, W., and Mulikidjanian, A. (1995) in *Photosynthesis: from Light to Biosphere* (Mathis, P., Ed.) pp 531–534, Kluwer Academic Publishers, Dordrecht.
55. Konermann, L., Yruela, I., and Holzwarth, A. R. (1997) *Biochemistry* 36, 7498–7502.
56. Jankowiak, R., Rätsep, M., Picorel, R., Seibert, M., and Small, G. J. (1999) *J. Phys. Chem. B* 103, 9759–9769.
57. Thompson, L. K., and Brudvig, G. W. (1988) *Biochemistry* 27, 6653–6658.
58. Stewart, D. H., and Brudvig, G. W. (1998) in *Photosynthesis: Mechanisms and Effects* (Garab, G., Ed.) pp 1113–1116, Kluwer Academic Publishers, Dordrecht.
59. Stewart, D. H., and Brudvig, G. W. (1998) *Biochim. Biophys. Acta* 1367, 63–87.

BI001246J